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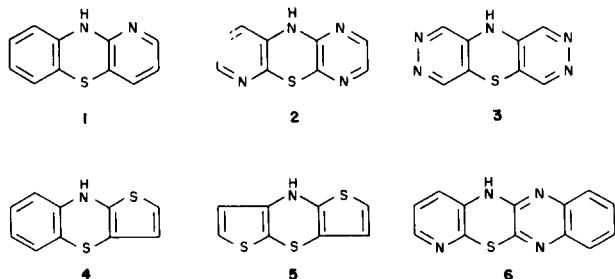
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This paper describes the unequivocal synthesis of three new heterocycles: 2,3,6-triazaphenothiazine (**7**) and 2,3,6,9-tetraazaphenothiazine (**18**), both parents of the respective ring systems and the 6,8-dihydro-7,9-dioxo derivative of 2,3,6,8-tetrahydrophenothiazine (**12**). These compounds were obtained by base-catalysed condensation of 4,5-dichloropyridazine (**8**) with the appropriate *o*-amino-heterocyclic thiols **9**, **11**, **16**. In contrast, reaction of 2,3,5-trichloropyridazine (**20**) with 4,6-diaminopyrimidine-5-thiol (**21**) and 2-amino-6-picoline-3-thiol (**28**) gave 5-chloro-2,3-bis(diaminopyrimidinyl-5-thio)pyrazine (**33**) and 5-chloro-2,3-bis(2-amino-6-picolinyl-3-thio)pyrazine (**29**), respectively. Chloropyridazine (**34**) and 3-aminopyridine-2(1*H*)-thione (**9**) followed the latter reaction path to 2-(3-amino-2-pyridyl) pyrazinyl sulfide (**35**) in good yields. Structural assignments were based on their infrared, ultraviolet, nmr and mass spectra.

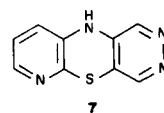
J. Heterocyclic Chem., **20**, 1047 (1983).

Previous reports have highlighted the importance of phenothiazine and its numerous derivatives in medicine, agriculture and industry [5,6]. New heterocyclic rings derived from phenothiazine were therefore sought in a search for improved agents in these areas. Among the major accomplishments is the replacement of one or more of the benzene rings in phenothiazine with heterocyclic systems, namely pyridine [7-10], pyridazine [12-16], pyrimidine [17-19], pyrazine [20-23], quinoxaline [24-25], 1,2,4-triazine [26], thiophene [27-28] and 1,4-thiazine [29]. These modifications have produced such interesting congeners of phenothiazine as 1-azaphenothiazine (**1**) [6], 1,4,6-triazaphenothiazine (**2**) [23], 2,3,7,8-tetraazaphenothiazine (**3**) [30], 9*H*-thieno[2,3-*b*]1,4]benzothiazine (**4**) [30], 4*H*-dithieno[2,3-*b*:2',3'-*e*]1,4]thiazine (**5**) [31-32] and 1,4,6-triazabenzob[*b*]phenothiazine (**6**) [33].

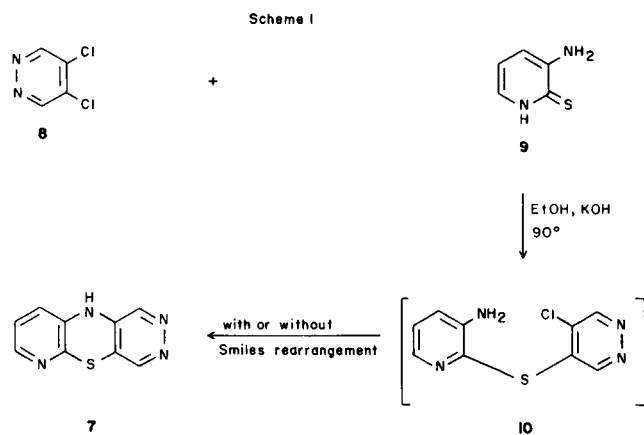


Although all of the derivatives of these classes of phenothiazine have not been fully evaluated, interesting antipsychotic [34-37] and CNS-depressant activities [38-40] have been recorded for a number of them. Currently some derivatives of **1** are in use as neuroleptic [41], antihistaminic [42], antiemetic [43] and antitussive [44] agents. The observation of these effects in the aza congeners of phenothiazine has led to the synthesis of a great variety of aza-

phenothiazines. In the more complex triaza- and tetraaza-phenothiazines, only four and seven [5,6] respectively, out of the possible 24 and 35 isomeric ring systems have been reported. In the preceding paper [45] we described the preparation of some derivatives of the 2,3,6-triazaphenothiazine ring system **7**. We have now prepared **7**, the parent heterocycle.



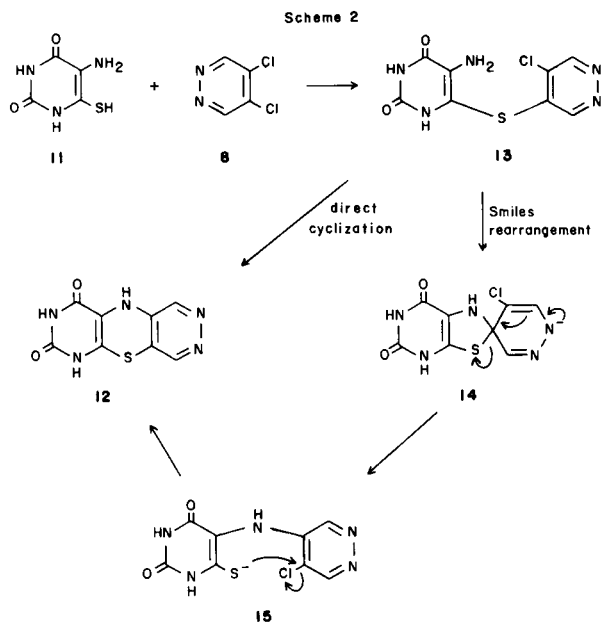
On heating a mixture of 4,5-dichloropyridazine (**8**) [30,46] and 3-aminopyridine-2(1*H*)-thione (**9**) in ethanol in the presence of potassium hydroxide at 90° for 10 minutes, 2,3,6-triazaphenothiazine (**7**) was isolated in 63% yield (Scheme 1). Microanalysis, infrared, ultraviolet and



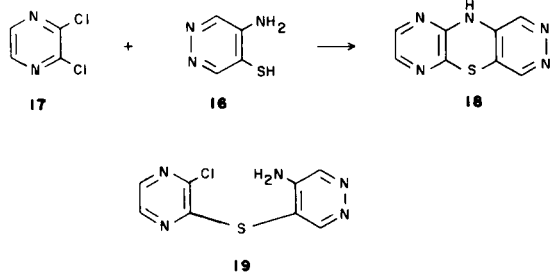
nmr spectroscopy agree with the assigned tricyclic structure. Since the condensation reaction was carried out in alkaline medium, the diaryl sulfide **10** was assumed to be

formed first in the reaction before the cyclization step. Cyclization of **10** then proceeds with or without Smiles rearrangement to yield the same product **7**.

A similar reaction of 4,5-dichloropyridazine (**8**) with 5-aminouracil-6-thiol (**11**) [47] gave 2,3,6,8-tetraazaphenothiazine-7,9(6*H*,8*H*)dione (**12**), the first member of the 2,3,6,8-tetraazaphenothiazine system. As in compound **7**, the same product is expected whether the reaction proceeds with or without Smiles rearrangement [5,48] of the intermediate pyrimidinylpyridazinyl sulfide **13** as shown in Scheme 2.



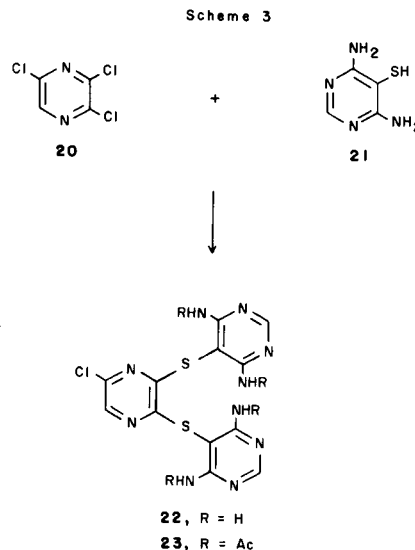
The reaction of a mixture of 4-aminopyridazine-5-thiol (**16**) [30,49] and 2,3-dichloropyrazine (**17**) [50-51] at room temperature in the presence of ethanolic potassium hydroxide led to 2,3,6,9-tetraazaphenothiazine (**18**) another new parent heterocycle in the tetraaza-series. Microana-



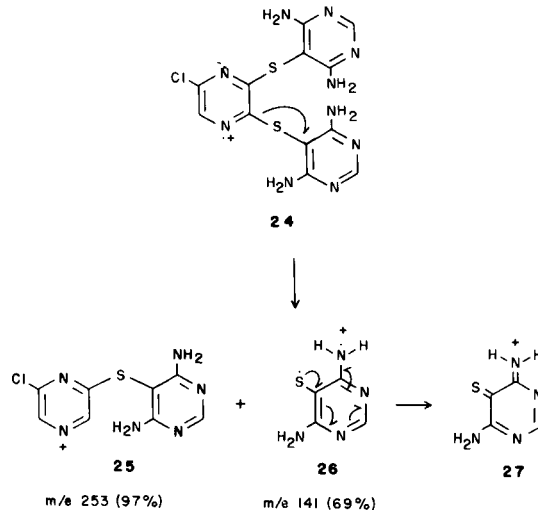
lysis, infrared and ultraviolet spectroscopy provide evidence for the assigned structure. Confirmatory proof of structure was obtained from the nmr spectrum which gave signals at δ 9.2 (broad, 10-NH), 9.26 (singlet, 1-H), 9.16 (singlet, 7-H and 8-H) and 9.01 (singlet, 4-H). As in compounds **7** and **12**, the same product is expected whether or

not the reaction proceeds by Smiles rearrangement of the sulfide **19**.

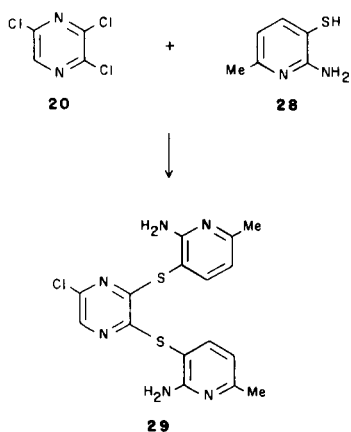
Base-catalysed reaction of 2,3,5-trichloropyrazine (**20**) [52] with 4,6-diaminopyrimidine-5-thiol (**21**) on the other hand gave, contrary to expectation, an 87% yield of 5-chloro-2,3-bis(4,6-diaminopyrimidinyl-5-thio)pyrazine (**22**, R = H). Structural assignment was based on microanalysis, infrared, ultraviolet and nmr spectra. Acetylation with acetic anhydride gave the tetraacetyl derivative **23**, R = Ac (Scheme 3).



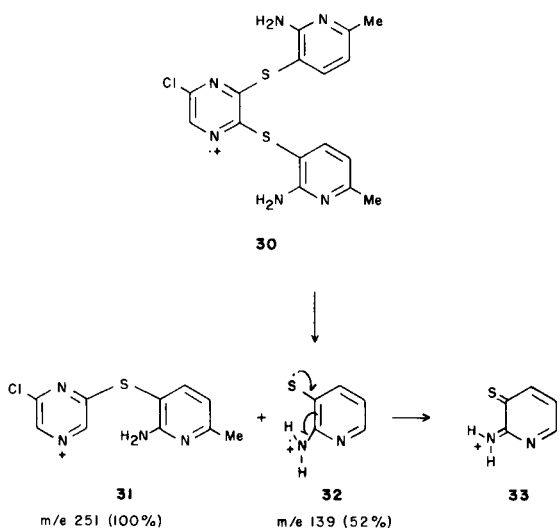
Confirmatory evidence of structure was obtained by mass spectroscopy which showed two prominent fragments at m/e 253 (97%) and m/e 141 (69%) due respectively to the pyrazinylpyrimidinyl sulfide ion **25** and 4,6-diaminopyrimidinyl-5-thiol diradical **26** which probably rearranges to give the radical ion **27**.



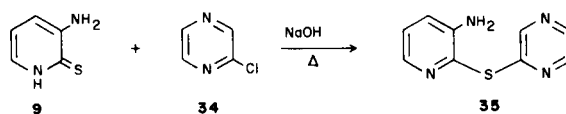
2,3,5-Trichloropyrazine (**20**) was also reacted with 2-amino-6-picoline-3-thiol (**28**) [53] under alkaline conditions to yield 5-chloro-2,3-bis(2-amino-6-picolinyl-3-thio)pyrazine



(29). The assigned structure was in agreement with micro-analytical and spectroscopic data. The fragmentation pattern in the mass spectrum of compound **29** was similar to that of compound **22**, R = H except that in the latter case the fragment ion **31**, formed by loss of one of the 2-amino-6-picolinyl-3-thiol radicals was the base peak in the spectrum. As a matter of fact, recognition of a major peak at *m/e* 139 has been used satisfactorily in this work in deciding whether phenothiazinoid or diaryl sulfides are formed by the reaction of compound **28** with *o*-dichloro heterocyclic compounds.



Such reactions leading to sulfides of types **24** and **29** are not limited to 2,3-dichloropyridazines. 2-chloropyridazine has sufficient reactivity to condense with *o*-amino-heterocyclic thiols. Thus when a mixture of 2-chloropyridazine **34** [51,54] and 3-aminopyridine-2(*H*)-thione (**9**) was refluxed for three hours in the presence of sodium hydroxide, 2-(3-amino-2-pyridyl)pyridazinyl sulfide (**35**) was obtained in good yield.



The assigned structure was in good agreement with analytical and spectroscopic data. This reaction confirms the ease of condensation of *o*-amino-heterocyclic thiols with halogenopyridazines.

EXPERIMENTAL

Melting points were determined with a Thomas-Hoover melting point apparatus and are uncorrected. The ir spectra were recorded on a Perkin-Elmer Model 337 grating infrared spectrophotometer using potassium bromide discs. The ultraviolet spectra were obtained with a Bausch and Lomb Spectronic 505 spectrometer using matched 1 cm quartz cells. The absorption maxima are reported in nanometers. The ¹H nmr spectra were obtained in the solvent indicated using a Varian A-60 or EM-360 spectrometer. Chemical shifts are reported in ppm from TMS used as an internal standard and are given in δ units. The following abbreviations were used to designate the multiplicity of individual signals: s = singlet, d = doublet and m = multiplet. Mass spectral data were recorded on a Hewlett Packard Model 5980 mass spectrometer at 70 eV. Elemental analyses were performed by MHW Laboratories, Phoenix, Arizona and Istituto Superiore di Sanita, Rome, Italy.

2,3,6-Triazaphenothiazine (7).

To a stirred solution of 4,5-dichloropyridazine (**8**, 0.15 g, 1 mmole) [30,46] in 15 ml of ethanol was added slowly a solution of 3-aminopyridine-2(*H*)-thione (**9**, 0.127 g, 1 mmole) [33] and potassium hydroxide (0.15 g) in ethanol (25 ml). The reaction mixture was heated on a steam bath at 90° (bath temperature) for 10 minutes and then allowed to stand at room temperature for an additional 14 hour period. During this time a precipitate was formed. The crude product was collected by filtration, washed with water and air-dried. Recrystallization of the precipitate from methanol-water (10:1) yielded cream-colored crystals of 2,3,6-triazaphenothiazine (**7**) (0.95 g, 63%), mp 214-215°; ir: ν max 3390-3345 cm^{-1} (Δ NH); uv (95% ethanol): λ max (ϵ) 254 (16,300), 283 (14,600); ¹H nmr (DMSO-*d*₆): δ 9.26 (s, 1-H), 9.06 (s, 4-H).

Anal. Calcd. for C₈H₆N₄S: C, 53.47; H, 2.99; N, 27.71. Found: C, 53.24; H, 3.16; N, 27.38.

2,3,6,8-Tetraazaphenothiazine-7,9(6*H*,8*H*)-dione (12).

To a stirred solution of 4,5-dichloropyridazine (**8**, 0.15 g, 1 mmole) in ethanol (15 ml) at room temperature was added slowly 5-aminouracil-6-thiol (**11**, 0.15 g, 1 mmole) [47] and potassium hydroxide in 25 ml of 50% ethanol. The reaction mixture was stirred for 24 hours at room temperature. The precipitate which formed was collected by filtration and washed with water. Recrystallization of the crude product from water furnished compound **12** (0.13 g, 50% yield), mp 263-265°; uv (water, saturated solution): λ max 258, 276 nm; ir: ν max 3340-3290 (broad, Δ NH), 1680 cm^{-1} (C=O); ¹H nmr (DMSO-*d*₆): δ 9.78 (s, 1-H), 9.03 (s, 4-H).

Anal. Calcd. for C₈H₆N₅O₂S: C, 40.86; H, 2.14; N, 29.78. Found: C, 41.06; H, 2.28; N, 30.17.

2,3,6,9-Tetraazaphenothiazine (18).

To a mixture of 4-aminopyridazine-5-thiol (**16**) (0.127 g, 1 mmole) and potassium hydroxide (0.25 g) in ethanol (25 ml) was added 2,3-dichloropyridazine (**17**, 0.15 g, 1 mmole) [50,51] in ethanol (10 ml) at room temperature. The mixture was allowed to react for 8 hours during which a precipitate formed. The crude product was collected and washed with cold water (20 ml) and air dried. Recrystallization from an ethanol-water mixture furnished 2,3,6,9-tetraazaphenothiazine (**18**) (0.07 g, 30% yield), mp

196-197°; ir: ν max 3380-3350 cm^{-1} (10-NH); uv (95% ethanol): λ max (ϵ) 273 (13,700), 278 (10,600); ^1H nmr (DMSO- d_6): δ 9.26 (s, 1-H), 9.20 (s, b, 10-NH), 9.16 (s, 7-H, 8-H), 9.01 (s, 4-H).

Anal. Calcd. for $\text{C}_8\text{H}_5\text{N}_5\text{S}$: C, 47.30; H, 2.48; N, 34.47. Found: C, 47.42; H, 2.13; N, 34.62.

5-Chloro-2,3-bis(4,6-diaminopyrimidinyl-5-thio)pyrazine (**22**, R = H).

4,6-Diamino-5-pyrimidinethiol (**21**, 2.84 g, 20 mmoles) was placed in the reaction flask to which was added 4 g of potassium hydroxide in water (15 ml) and *N,N*-dimethylacetamide (DMAC) (30 ml). The mixture was warmed to dissolve compound **21** completely and then 2,3,5-trichloropyrazine (**20**, 4.58 g, 25 mmoles) was added. The reaction mixture was heated under reflux for 4 hours. A yellow precipitate was formed as soon as refluxing started and it persisted throughout the reflux period. The mixture was then poured onto crushed ice (500 g), cooled further and filtered. The impure product was recrystallised from aqueous DMAC after treatment with activated charcoal to afford 5-chloro-2,3-bis(4,6-diaminopyrimidinyl-5-thio)pyrazine (**22**, R = H) (7.34 g, 93% yield) as yellow-green plates, mp $> 300^\circ$; uv (methanol): λ max (ϵ) 251 (8,817), 323 (6,056); ir: ν max 3430, 3330, 3160, 1650, 1616, 1575, 1534, 1490, 1474, 1366, 1347, 1310, 1290, 1202, 1184, 1160, 1130, 1090, 1038, 1010, 990, 962, 890, 867, 843, 777, 735, 694, 640 cm^{-1} ; ^1H nmr (DMSO- d_6): δ 8.30 (s, 2'-H, 2''-H), 7.90 (s, 6-H), 6.57 (s, b, 4'-NH₂, 4''-NH₂, 6'-NH₂, 6''-NH₂); ms: m/e (relative intensity) 77 (12), 82 (13), 83 (13), 85 (30), 86 (48), 87 (39), 88 (17), 91 (15), 95 (30), 97 (26), 114 (30), 141 (69), 147 (16), 149 (11), 219 (20), 221 (10), 252 (11), 253 (97), 254 (20), 255 (35), 288 (38), 290 (26), 394 (M⁺, 100%), 395 (M + 1, 14), 396 (M + 2, 36).

Anal. Calcd. for $\text{C}_{12}\text{H}_{11}\text{ClN}_{10}\text{S}_2$: C, 36.50; H, 2.79; N, 35.49; Cl, 9.00; S, 16.23. Found: C, 36.48; H, 2.55; N, 35.75; Cl, 9.02; S, 16.11.

5-Chloro-2,3-bis(4,6-diacetamidopyrimidinyl-5-thio)pyrazine (**23**, R = Ac).

5-Chloro-2,3-bis(4,6-diaminopyrimidinyl-5-thio)pyrazine (**22**, R = H) (3.95 g, 10 mmoles) was placed in a reaction flask containing acetic anhydride (50 ml) and pyridine (1 ml). The mixture was heated at reflux temperature for 2 hours in an oil bath. The clear solution was treated with activated charcoal, filtered and cooled to ice-bath temperature. Crushed ice was then added and the mixture gently warmed until a homogeneous solution was achieved. The yellow solution was made alkaline with concentrated ammonia while cooling. The greenish yellow solid that precipitated out was collected by filtration and recrystallized from *N,N*-dimethylacetamide to yield 5-chloro-2,3-bis(4,6-diacetamidopyrimidinyl-5-thio)pyrazine (**23**, R = Ac) (4.95 g, 88% yield) as yellowish-green plates, mp $> 300^\circ$; uv (methanol): λ max (ϵ) λ infl 235 (35,100), 275 (5,850), 372 (11,700); ir: ν max 3200, 3000, 2920, 2840, 1702, 1680, 1585, 1557, 1520, 1484, 1407, 1360, 1300, 1250, 1232, 1191, 1165, 1097, 1055, 1039, 1000, 953, 904, 865, 820, 770, 733, 672 and 642 cm^{-1} ; ^1H nmr (DMSO- d_6): δ 10.72 (s, b, 4'-NH₂, 4''-NH₂, 6'-NH, 6''-NH₂), 8.17 (m, 2'-H, 2''-H, 6-H), 2.10 (s, four CH₃ protons).

Anal. Calcd. for $\text{C}_{28}\text{H}_{19}\text{ClN}_{10}\text{O}_4\text{S}_2$: C, 42.67; H, 3.38; N, 24.89; Cl, 6.31; S, 11.38. Found: C, 42.88; H, 3.20; N, 25.00; Cl, 6.29; S, 11.24.

5-Chloro-2,3-bis(2-amino-6-picolinyl-3-thio)pyrazine (**29**).

2-Amino-6-picoline-3-thiol (**28**) (5.6 g, 40 mmoles) was placed in a 250 ml three-necked flask equipped with a dropping funnel, a mechanical stirrer and a reflux condenser. Sodium hydroxide (6 g) in water (50 ml) was added and the mixture warmed to dissolve. *N,N*-Dimethylacetamide (30 ml) was then added followed by the addition of 2,3,5-trichloropyrazine (**20**, 9.18 g, 50 mmoles). The mixture was heated under reflux for 5 hours. At the end of the reflux period, the mixture was cooled and poured into a beaker containing ice-cold water (500 ml), stirred and cooled overnight in a refrigerator. The crude product was collected by filtration and recrystallized from a methanol-*N,N*-dimethylacetamide mixture after treating with activated charcoal to afford glistening golden-yellow needles of 5-chloro-2,3-bis(2-amino-6-picolinyl-3-thio)pyrazine (**29**) (12.81 g, 82% yield), mp 185-186°; uv (methanol): λ max (ϵ), 310 (5,940), 260 (14,850); ir: ν max 3455, 3290, 1660, 1635, 1590, 1565, 1530, 1500, 1470,

1430, 1405, 1380, 1365, 1344, 1333, 1290, 1205, 1190, 1162, 1130, 1080, 1042, 1006, 964, 933, 903, 842, 820, 745, 720 and 665 cm^{-1} ; ^1H nmr (DMSO- d_6): δ 8.30 (s, b, 2'-NH₂, 2''-NH₂), 7.40 (d, J = 8.2 Hz, 4'-H, 4''-H), 6.38 (m, 5'-H, 5''-H, 6-H), 2.27 (s, 6'-CH₃, 6''-CH₃); ms: m/e (relative intensity) 95 (18), 139 (52), 218 (15), 237 (9), 250 (21), 251 (100), 252 (21), 253 (34), 271 (17), 273 (9), 286 (49), 288 (34), 390 (M⁺, 79%), 392 (M + 2, 36).

Anal. Calcd. for $\text{C}_{16}\text{H}_9\text{ClN}_6\text{S}_2$: C, 49.17; H, 3.84; N, 21.51; Cl, 9.09; S, 16.39. Found: C, 49.12; H, 3.88; N, 21.25; Cl, 8.84; S, 16.41.

2-(3-Amino-2-pyridyl)pyrazinyl Sulfide (**35**).

3-Aminopyridine-2(1*H*)-thione (**9**) (3.15 g, 25 mmoles) was dissolved in a solution of sodium hydroxide (3 g) in water (30 ml). 2-chloropyrazine (**34**, 3.44 g, 30 mmoles) in propylene glycol (80 ml) was then added and the mixture heated under reflux for 8 hours. It was poured into a beaker containing crushed ice (600 g) and cooled further for 2 hours. The crude product which precipitated was collected by filtration and recrystallized from ethanol to afford after treatment with activated charcoal yellow needles of compound **35** (3.67 g, 72% yield), mp 212-213°; uv (methanol): λ max (ϵ) 235 (12,730), 300 (9,155), 348 (13,040); ir: ν max 3300-3200 (3-NH₂); ^1H nmr (DMSO- d_6): δ 9.10 (s, 3-H), 8.45 (s, 5-H, 6-H), 8.08 (s, b, 3'-NH₂), 7.00 (m, 4'-H, 5'-H, 6'-H); ms: m/e (relative intensity) 92 (17), 103 (17), 136 (13), 137 (10), 149 (10), 150 (16), 171 (63), 176 (23), 203 (36), 240 (M⁺, 100%), 245 (22), 246 (7).

Anal. Calcd. for $\text{C}_9\text{H}_8\text{N}_4\text{S}$: C, 52.94; H, 3.92; N, 27.45; S, 15.69. Found: C, 52.89; H, 3.72; N, 27.46; S, 15.78.

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